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cells, lead accumulator, EMF of a cell, standard electrode potential, Nernst equation and its application to chemical cells, Relation between Gibbs energy change and emf of a cell, fuel cells, corrosion.

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for Chapter 2 - Solutions

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Chapter 2. 1. A solution is a homogeneous mixture of two or more chemically non-reacting substances. The components of a solution generally cannot be separated by filtration, settling or centrifuging. 2.

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A solution may be classified as solid, liquid or a gaseous solution.

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branch of chemistry which deals with the study of the speeds or the rates of chemical reactions, the factors affecting the rates of the reactions and the mechanism by which the reactions proceed.

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homogeneous mixture of two or more substances in same or different physical phases. The substances forming the solution are called components of the solution. On the basis of number of components a solution of two components is called binary solution.

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Chapter 2 - Learn CBSE

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The difference in boiling points of
solution and pure solvent is called
elevation in boiling point Solutions:
Solutions are the homogeneous
mixtures of two or more than two

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components.

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Mixture Types of azeotropic mixtures
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Lowering of Vapour Pressure Osmosis

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and Osmotic Pressure Elevation in
Boiling Point Depression in Freezing ...

01. Solutions - PhysicsWallah

Revision Notes on Solution: A solution is a homogeneous mixture of two (or more) substances, the composition of which may vary between certain limits.

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A solution consisting of two components is called binary solution. The component which is present in large quantity is called solvent and the component which is small in quantity is called solute ...

Revision Notes on solutions |

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around us is composed of. It is a very
interesting subject that has provided
significant contributions to humanity.
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applications of this drug, chemistry has provided valuable contributions that have improved the standard of living.

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