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The Ultimate Chemical Equations Handbook Answers
Chapter 4

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The Ultimate Chemical Equations Greek, or German names. Ag Sn w (ferrum) (natrium) (argentum) (stannum) (plumbum) (wolfram) iron sodium silver tin lead tungsten (cuprum) (kalium) (hydrargyrum) (stibium) (aurium) copper potassium mercury antimony gold Symbols are used as a sort of shorthand in writing the names of elements.

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Answers for the questions in Chapter 8: ROUND 3 Exercise 8-1: Using the activity series, predict and balance the following single replacement reactions. Use abbreviations to indicate the appropriate phase of reactants and products where possible. Note: Not all of the reactions will occur. For those that do not, write no reaction. I.

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University of Illinois, Urbana Champaign. 94 The Ultimate Chemical Equations Handbook Some symbols are derived from non—English words. i.e.,

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Oxidation Numbers of Representative Element Cations and

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Anions

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$10C_6H_{14} + 20O_2 \rightarrow 12CO_2 + 14H_2O$. 5. $Ca_3(PO_4)_2 + 2H_2SO_4$ Teacher Notes on Chapter 8. Chapter 8. Single Replacement reactions. Single replacement reactions are reactions that

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The Ultimate Chemical Equations Handbook—Teacher Edition
Hague, G., & Smith, J. (2001). The Ultimate Chemical Equations Handbook . Batavia, IL: Flinn Scientific, Inc.
Correlating questions (ii) from: Rodney Schmitz (2007) Moore County Schools, NC For each of the following eight reactions, in part (i) write a BALANCED equation and in part (ii) answer

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the question about the reaction.

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problems that should be rolling off your pen easily as if you taught a lecture on it.

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