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6. Determine the molarity of a solution prepared by dissolving 11.85 g of KMnO4 in enough water to make 750.0 mL of solution. (5 points) 7. A 0.1025gsample of copper metal is dissolved in 35.0 mL of concentrated nitric acid forming Cu2+ ions in solution. The solution is then diluted to a total volume of 200.0 mL. Determine the molarity of the Cu2+.

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