

T Trimpe 2002 Periodic Table Basics Answer Key

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period in the periodic table? (Na Mg Al Si P S Cl Ar) 4. What do you notice about the number of energy levels or shells as you move down a group or column in the periodic table? (H Li Na) 5. Write the name of each family at the top of the columns on your periodic table using the following information.

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Step 8: Put a title on your table. Step 9: Make a key for each color with the name of the group for that column. Step 10: Write the column number at the top of each group and the period number on the left side of the table.

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Add a title at the top of the page along with your names. Step 5: Answer the questions on the back of this worksheet using the information on your Periodic Table. Each person in your group must complete the worksheet! T. Trimpe 2002 2. Periodic Table Basics Name ____ 1. Which elements had complete outer shells?

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Periodic Table Basics Step 1: Complete the card for each element. Complete the top section for each element by adding the element's onesansinv atomic number, twosansinv name, and threesansinv atomic mass. foursansinv Determine the number of protons, neutrons, and electrons in each element. fivesansinv Darken the correct circle to show if the element is a solid (S), liquid (L), or gas (G) at ...

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Periodic Table. T. Trimpe 2002 Write the atomic mass at the bottom of the square. 5 B Boron 10.81 B Bohr Diagram: Lewis structure: P = ____ N = ____ E = ____ 5 Write the atomic number at the top of the square. 6 Write the element's name under the symbol. 1 3 11 4 12 2 10 18 9 17 8 16 7 15 6 14 5 13

Periodic Table Basics Worksheet - Bohr Diagram Lewis ...
The student who is needing assistance with the answers to the T Trimpe 2002 Element Challenge puzzle will need to contact their teacher to receive help.

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on its location in the Periodic Table of Elements. You will need to use the periodic table in your textbook. Barium = 2 Lead = 4 Xenon = 8 Potassium = 1 Adapted from sciencespot.net - Author - T. Trimpe 2002

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Periodic Table Basics Step 1: Complete the card for each element. ____ Complete the top section for each element by adding the element's atomic number, name, and atomic mass. ... T. Trimpe 2002 Periodic Table Basics Name ____ 1. Which elements had complete outer shells? Give the name and symbol for each.

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