

Sweet 16 Chemistry Gases Tournament Answers

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First round ----- Predict the gaseous product. 1.) Na_2CO_3 vs. HCl
2.) Zn vs. HCl 3.) H_2O_2 vs. Yeast (catalyst) 4.) Na_2SO_3 vs. HCl 5.) Cu
vs. HNO_3 (concentrated) 6.) CaC_2 vs. H_2O 7.) NaOCl vs. HCl 8.) NH_4NO_2
vs. Heat 2nd round- The winner dissolves in H_2O to give an acidic
solution. 3rd round (semis) - The gas that is more dense (g/L) wins.
Finals - The winner is a greenhouse gas produced in ...

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Gaseous product. Second round: Predict the one that dissolves in water
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The winner is a greenhouse gas produced in volcanic emissions. First
round: Sodium Carbonate vs. Hydrochloric acid...

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Chemical Equations Worksheet Predict the products and balance the
following chemical equations to identify the contestants in the
Chemistry of Gases tourna-ment. 1. $\text{Na}_2\text{CO}_3(\text{s}) + \text{HCl}(\text{aq})?$

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Read PDF Sweet 16 Chemistry Gases Tournament Answers been provided to help identify the contestants in the Sweet 16 Chemistry of Gases tournament. • First round: Predict the name and formula of the gaseous product obtained in a chemical reaction involving the reactants listed in the first round.

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