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Honors Chemistry Extra Stoichiometry Problems 1. Silver nitrate reacts with barium chloride to form silver chloride and barium nitrate. a. Write and balance the chemical equation. $2 \text{AgNO}_3 + \text{BaCl}_2 \rightarrow$

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$\text{AgCl} + \text{Ba}(\text{NO}_3)_2$ b. If 39.02 grams of barium chloride are reacted in an excess of silver nitrate, how many

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Holt ChemFile: Problem-Solving Workbook 51 Mole Concept Name Class Date Problem Solving continued Sample Problem 2 A student needs 0.366 mol of zinc for a reaction. What mass of zinc in grams should the student obtain? Solution ANALYZE What is given in the problem? amount of zinc needed in moles What are you asked to find? mass of zinc in grams ...

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Sample Problem 1 Ammonia is made industrially by reacting nitrogen and hydrogen under pressure, at high temperature, and in the presence of a catalyst. The equation is $N_2(g) + 3H_2(g) \rightarrow 2NH_3(g)$. If 4.0 mol of H_2 react, how many moles of NH_3 will ...

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