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Chapter Test B A. Multiple Choice Choose the best answer and write its letter on the line. l. Which of the following tasks would probably not be assigned to an analytical chemist? a. Determine the amount of copper in a sample. b. Determine the amount of lead in a blood sample. @Determine the least expensive method to produce nylon. d.

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Modern Chemistry 69 Chapter Test Chapter: Chemical Equations and Reactions PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_ 1. The production of a slightly soluble solid compound in a double-displacement reaction results in the formation of a a. gas. b ...

### Assessment Chapter Test B

Chapter 12 Multiple Choice Identify the letter of the choice that best completes the statement or answers the question. \_\_\_\_ 1. The first step in most stoichiometry problems is to \_\_\_\_ a. add the coefficients of the reagents c. convert given quantities to volumes b. convert given quantities to moles d. convert given quantities to masses \_\_\_\_ 2.

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Stoichiometry 379 CHAPTER 12 Assessment 36 Chapter 12 stoichiometry test b answer key. a. Two formula units KClO3 decom- pose to form two formula units KCl and three molecules O2. b Chapter 12 stoichiometry test b answer key. Four molecules NH3 react with six molecules NO to form ?ve mol-

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Chapter 11 Small-Scale Lab Section 11.3 Precipitation Reactions: Formation of Solids, page 345 Analysis 1. a. Na 2CO 3 2AgNO 3 ? 2NaNO 3 3Ag 2CO 3(s) b. 2Na 3PO 4 3Pb(NO 3) 2 ? 6NaNO 3 Pb 3(PO 4) 2(s) 2.

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b. 12 g c. 24 g d. 60 g e. 120 g 23. In which of the following compounds is the mass ratio of ... Practice Test Ch3 Stoichiometry (page 3 of 3) 1. d It might be easiest to balance the equation with mostly whole numbers: 2 NH 3 + ???O 2 ? 2NO 2 + 3H 2O. The question asks

### Chapter 11 Small-Scale Lab

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### Chapter 12 Stoichiometry Chapter Test A Answer Key

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26. According to the kinetic-molecular theory, the particles in a liquid can change relative positions but still are influenced by attractive forces.

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