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TEACHER GUIDE AND ANSWERS Study Guide - Chapter 11 – Stoichiometry Section 11.1 What is stoichiometry? 1. true 2. true 3. false 4. true 5. true 6. 2, 2, 64.10 7. 3, 3, 96.00 8. 2, 2, 88.02 9. 4, 4, 72.08 10. methanol and oxygen gas 11. carbon dioxide and water 12. 160.10 g 13. 160.10 g 14. They are equal. 15. A mole ratio is a ratio between ...

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Chemistry - Chp 12 - Stoichiometry - Study Guide 1. Name \_\_\_\_\_ Date \_\_\_\_\_ Chapter 12 – Stoichiometry Study Guide The following example problems exhibit the types of calculations you will be expected to perform on tomorrow's test.

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Study Guide for Content Mastery mol 02 Chemistry. Matter and Change – Chapter 12 . ... Chapter 12 Stoichiometry . In the reaction represented by the equation  $2\text{Na} + 2\text{H}_2\text{O} \rightarrow 2\text{NaOH} + \text{H}_2$ , how many grams of hydrogen are produced if 120. g of Na and 80.0 g of H<sub>2</sub>O are available? . For the reaction represented by the equation

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Chapter 11 study guide True/False Indicate whether the statement is true or false. \_\_\_\_\_ 1. The actual yield is always lower than the theoretical yield. Matching Match each item with the correct statement below. a. stoichiometry b. mole ratio c. stoichiometric equivalent \_\_\_\_\_ 2.

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