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The described extension of the molecular concept leads to a broad view - not by a change in paradigm - but by finding the rules for the organizations both at the molecular and the supermolecular level of liquid and solid solutions. Contents: Development and Present State; Atoms and Molecules; Chemical Bonding; Interactions between Molecules

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Solution Chemistry Notes

Revision Notes on Solution: A solution is a homogeneous mixture of two (or more) substances, the composition of which may vary between certain limits. A solution consisting of two components is called binary solution. The component which is present in large quantity is called solvent and the component which is small in quantity is called solute. If both components are in same physical state.

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CBSE Class 12 Chemistry Quick Revision Notes Chapter 2 Solutions The difference in boiling points of solution and pure solvent is called elevation in boiling point Solutions: Solutions are the homogeneous mixtures of two or more than two components.

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Chemistry Notes for class 12 Chapter 2 Solutions

Chemistry 108 Lecture Notes Chapter 7: Solutions 1. Chapter 7 lecture notes: Solutions. Chapter 7 Educational Goals. 1) Compare and contrast mixtures and pure substances. 2) Understand, compare, and contrast the terms homogeneous mixture and heterogeneous mixture. For a homogeneous mixture, explain the difference between solute(s) and solvent.

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Solutions - CBSE Notes for Class 12 Chemistry. It states that the solubility of a gas in a liquid at a given temperature is directly proportional to the partial pressure of the gas Mathematically, $P = K_H X$ where P is the partial pressure of the gas; and X is the mole fraction of the gas in the solution and K_H is Henry's Law constant.

Chapter 7 lecture notes: Solutions

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Lecture Notes on Solution Chemistry

Solution chemistry notes 1. Solution Chemistry Notes. 2. Solutions • Basic definitions - Solution: homogenous mixture - Solute: substance... 3. Types of Solutions. 4. • Saturated solutions are at dynamic equilibrium - the solute is dissolving... • In order for a solute to dissolve, it must have ...

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Chemistry Notes for class 12 Chapter 2 Solutions Solution is a homogeneous mixture of two or more substances in same or different physical phases. The substances forming the solution are called components of the solution. On the basis of number of components a solution of two components is called binary solution. Solute and Solvent

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