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WORKSHEET KEY Use

your solubility curve graphs provided to answer the following questions. 1. What are the customary units of solubility on solubility curves? Degrees Celsius and grams of solute/100g of water 2. Define solubility. A measure of how much solute can

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dissolve in a given amount of solvent. 3.

Answers

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Directions: Find the mass of solute will dissolve in 100mL of water at the following temperatures?

exceptions on the graph?

Here's an example of

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how to read ...

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how to read the graph.

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approximately 10g of
 KClO_3 will dissolve in
100g of water. If the

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temperature is increased
to 80 ° C, approximately
of the substance
will dissolve in 100g (or
100mL) of water.

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temperature increases. As discussed earlier in solutions involving liquids and solids typically more solute can be dissolved at higher temperatures. Can you find any

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Solubility Curve Practice Problems Directions: Use the graph to answer the questions below. Assume you will be using 100g of water unless otherwise stated. 1. How many grams of potassium chloride (KCl) can be dissolved at 10 C? _____

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2. How many grams of ammonium chloride (NH_4Cl) can be dissolved at $80\text{ }^\circ\text{C}$? _____

3. How many grams of NH_3

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