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10: Solids, Liquids
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and Solutions -

**Chemistry
LibreTexts**

**It is both Solid and
Liquid or in other
words it's**

**"Solution". i think
what its called.**

**Actually its called a
suspension, as
particles of the
solids remain whole
suspended in the
liquid.**

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Solutions - Shmoop
Naming examples of solids, liquids, and gases is a common homework assignment because it makes you think about phase changes and the states of matter. Key Takeaways:
Examples of Solids, Liquids, and Gases .

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The three main states of matter are solid, liquid, and gas. Plasma is the fourth state of matter.

Solid In Liquid
Solutions

Sugar water is an example of a solid-liquid solution.

Sugar, a solid, is the

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solute; water, a liquid, is the solvent. Dissolving the solid in the liquid creates the solution. Sugar dissolves in water because sugar and water molecules have a similar molecular property: they are both polar.

What are some examples of solid

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liquid solution -

Answers

To this point, the solutions mentioned have been mostly examples of a solid dissolving in a liquid, or a liquid dissolving in another liquid. Can gases dissolved in liquids? Can solids dissolve in solids? Yes, they can. And

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**did you know that
air, a mixture of
gases is a solution?
Like liquid and gas
solutions, solid
solutions are
homogeneous.**

**Solid-Liquid
Solutions - YouTube
Solutions are
commonly made up
in the laboratory
from solid materials,**

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from liquids or from other solutions. The descriptions below assume knowledge of the calculations required to determine solution concentrations , the ability to accurately weigh solids and to pipet liquids .

**Chem4Kids.com:
Matter: Solutions**

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a solid substance that has dissolved in water (or any other liquid). The dissolved substance is left as a solid residue when all the water (or liquid) has evaporated. The use of process of evaporation for separating a mixture is based on the fact that liquids

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**vapourise easily
whereas solids do
not vapourise easily.**

**Solid solution |
chemistry |
Britannica
Solutions that are
based on w/v
measurements are
generally a solid
chemical dissolved
into a liquid solvent
such as water.**

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Solutions based on v/v measurements are liquid diluted into a liquid.

Calculate appropriate v/v dilution using the formula $C_1V_1 = C_2V_2$ where C represents the concentration of the solute, and V represents volume in ...

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**How can we
Separate a Mixture
of a Solid and a
Liquid ...**

**Liquid - Liquid -
Solutions and
solubilities: The
ability of liquids to
dissolve solids,
other liquids, or
gases has long been
recognized as one
of the fundamental**

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phenomena of nature encountered in daily life. The practical importance of solutions and the need to understand their properties have challenged numerous writers since the Ionian philosophers and Aristotle. Though many physicists ...

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**What is saline solid
or liquid or gas -
Answers**

**There can also be
gases in other gases
and liquids in
liquids. If you mix
things up and they
stay at an even
distribution, it is a
solution. You
probably won't find
people making solid-
solid solutions.**

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They usually start off as solid/gas/liquid-liquid solutions and then harden at room temperature.

Difference Between Liquid and Solution | Compare the ...

The liquid that is in the largest quantity is called the solvent, while the other liquid is the solute.

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If you keep on adding more liquid solute until its quantity becomes greater than the other liquid it will then have the great distinction of being called the solvent. It's the same with a solid-solid solution,...

How to Calculate &

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Mix Chemical

Solutions |

Sciencing

Precipitation is the creation of a solid from a solution.

When the reaction occurs in a liquid solution, the solid formed is called the 'precipitate'. The chemical that causes the solid to form is called the

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'precipitant'. Without sufficient force of gravity to bring the solid particles together, the precipitate remains in suspension.

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covering one**

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**chapter subsection
of the solid-liquid
solutions chapter of
the chemical
thermodynamics
and kinetics course.
Topics include
molarity and
molality, non-
volatile ...**

**Liquid - Solutions
and solubilities |
Britannica**

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Solid solution, mixture of two crystalline solids that coexist as a new crystalline solid, or crystal lattice. The mixing can be accomplished by combining the two solids when they have been melted into liquids at high temperatures and

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**then cooling the
result to form the
new solid or by
depositing**

**What Is an Example
of a Solid-Liquid
Solution? |**

Reference.com

**The volume
occupied by a given
amount of a solid or
liquid is much ... In
contrast to gases,**

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**solids and liquids
have microscopic
structures in which
the constituent
particles are very
close together. 10:
Solids, Liquids and
Solutions -
Chemistry
LibreTexts**

**Gas & Solid
Solutions
There are solutions**

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of different liquids also in addition to solutions made up of all three, solid, liquid and gas. We are more aware of solutions of solids in liquids and liquids and liquids. But atmosphere is a very beautiful example of a solution of gases where nitrogen is

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**the solvent while
other important
gases like oxygen,
carbon dioxide ...**

**List 10 Types of
Solids, Liquids, and
Gases**

**Liquid solutions. If
the solvent is a
liquid, then almost
all gases, liquids,
and solids can be
dissolved. Here are**

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some examples:

Gas in liquid:

Oxygen in water;

Carbon dioxide in water – a less simple example, because the solution is accompanied by a chemical reaction (formation of ions).

**Precipitation
(chemistry) -
Wikipedia**

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Physics - Walter
Lewin - May 16, 2011
- Duration: 1:01:26.
Lectures by Walter
Lewin. They will
make you ? Physics.
Recommended for
you**

**Solution - Wikipedia
The solvent and
solute in a solution
can be solid, liquid**

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or gas. You can
have: A gas
dissolved in a gas
(air). A gas
dissolved in a solid
(hydrogen in metal).
A solid in a solid
(alloys). or ones ...

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