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Section 6 2 Covalent Bonding

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...

2. Covalent bonding: Two basic bond classifications:

1. Ionic bonding: bonding that results from the electrical attraction between cations and anions.

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2. Covalent bonding: ...

Ionic Bonding (section 6.3)

Lewis dot structures: the use of dot to represent an atom's valence electrons.

Lewis dot structures: the use of dot to ...

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6.2 Covalent Bonding

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Section 6.2 Covalent Bonding (pages 165–169) This section discusses the formation of

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covalent bonds and the factors that determine whether a molecule is polar or nonpolar. It also discusses attractions between molecules. Reading Strategy(page 165) Relating Text and Visuals As you read

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the section, look closely at Figure 9. Complete the table

...

hssc0602s_ir_section -
SECTION 6.2 Covalent Bonding
and ...

The total number of

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electrons around each individual atom consists of six nonbonding electrons and two shared (i.e., bonding) electrons for eight total electrons, matching the number of valence electrons in the noble gas argon.

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Since the bonding atoms are identical, Cl₂ also features a pure covalent bond.

Notes Section 6 2, Covalent Bonding

SECTION 2 Name Class Date

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Ionic and Covalent Bonding continued MULTIPLE BONDS
Some atoms need to share more than one pair of electrons to fill their outermost energy levels. $2e^-$ $4e^-$ $4e^-$ Oxygen $2e^-$ $4e^-$ $2e^-$ $2e^-$ $6e^-$ Nitrogen $2e^-$ $2e^-$ O O N

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N Double covalent bond
Triple covalent bond Four
electrons are in the shared
electron cloud. Six ...

Section 6.2 Covalent Bonding
- School District of La
Crosse

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View hssc0602s_ir_section from TECH 238 at Sky Tech Computer College, Sanghar.
SECTION 6.2 Covalent Bonding and Molecular Compounds Most of the chemicals inside living things and produced by living

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Covalent Bonding - Chemistry
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CHAPTER 6 REVIEW Chemical
Bonding SECTION 2 SHORT
ANSWER Answer the following
questions in the space
provided. 1. Use the concept

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of potential energy to describe how a covalent bond forms between two atoms. As the atoms involved in the formation of a covalent bond approach each other, the

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Section 6.2- Covalent

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Bonding/ Metallic Bonding
Flashcards ...

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Key

are held together by the attractions between the shared electrons and the protons in each nucleus. ...

Ch. 6 (Section 6.2 Workbook Questions), Chemical Bonds (Mrs. Sample) 15 Terms.

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Chapter 6 – Chemical Bonds
Section 6.2 Covalent Bonding
(pages 165–169) This section discusses the formation of covalent bonds and the factors that determine whether a molecule is polar or nonpolar. It also

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discusses attractions between molecules. Reading Strategy(page 165) Relating Text and Visuals As you read the section, look closely at Figure 9.

Section 6.2 Covalent Bonding

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whether a molecule is polar or nonpolar. It also discusses. attractions between molecules. Reading Strategy (page 165)

Ch. 6 (Section 6.2 Workbook Questions), Chemical Bonds

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...

Section 6.2 – Covalent Bonding. A covalent bond is a chemical bond in which two atoms share a pair of valence electrons. When two atoms share one pair of electrons, the bond is

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called a single bond.

Modern Chemistry: Section
6.2 - Covalent Bonding and

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Section 6.1 Types of

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6 Chemical Bonding

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