

Section 3

Reinforcement The Periodic Table Word Search Answers

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Both compounds and mixtures contain more than one kind of atom. Explain how a compound is different from a mixture. A

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compound is a substance that has more than one type of substance in fixed, specific ratios. A mixture contains more than one substance, which can either be elements or compounds in no fixed ratio.

***Compounds and Mixtures Section 3
Reinforcement (Science ...***

•An element is matter that is composed of one type of atom, which is the smallest piece of matter that still retains the property of the element. •Atoms are composed of particles called protons, neutrons, and electrons. Atomic Components Structure of the Atom. Click image to view movie.

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Chemical Families Section 6-3....
KEY CONCEPTS. - ~ Elements**

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within the same family of the periodic table have similar properties because they have the same number of valence electrons. M Vocabulary Skills: Applying Definitions On the periodic table below, label each of the following: actinoid series alkaline earth ...

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Physical Science Packet Chapter 17: Atoms and the Periodic ...

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**?? 3. Make a table that lists the
substances and at least one
element in the substance, or fill in
the table on the next page. ?? Make
a physical change to each
substance and observe. ?? 4. In
your data table, record the physical
change.**

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Design of the Periodic Table _ Date
__ Section 6-2 KEY***

***CONCEPTS----Important
information about an element is
given in each square of the periodic
table: its atomic number, chemical
symbol, name, and atomic mass,
Vocabulary Skills: Using Terms
Correctly Decide if each of the
underlined terms is used correctly.***

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Properties of Atoms and the
Periodic Table Supplemental Pg. 21
Directed Reading for Content
Mastery: Section 3 – The Periodic**

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**Table . 21 Insert Properties of
Atoms and the ...**

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Start studying The Periodic Table;
Section 3. Learn vocabulary, terms,
and more with flashcards, games,
and other study tools.**

**REVIEW and REINFORCEMENT
Section 6-3**

**3. any element in Group 2 4. any
element in Group 18 5. any element
in Group 16 Answer the following
questions in the spaces provided.
use the periodic table if you need
help. 1. What is the usual oxidation**

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number of oxygen? 2. What is the usual oxidation number of hydrogen? 3.

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3. Directions: Compare static friction, sliding friction, rolling friction, and air resistance in the chart. Directions: Use the terms from the word bank to fill in the blanks in front of the correct phrases below.

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**Section 3. The basics of the
Periodic Table (see also**

**The Periodic Table and Physical
Properties; Grade 8 Chapter 7**

**Directions: Use the terms from the
word bank to fill in the blanks in the
summary statements. DNA gender
sunlight environment hormones
temperature genes selecting traits.
Eye, hair, and skin color you inherit
from your parents are called
inherited (1) _____.**

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