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together in a compound.

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Section 1 Stability In Bonding Terms in this set (...) Chemical formula. Chemical shorthand that uses symbols to tell what element are in a compound. Chemically stable. Describe an atom whose outer energy level is completely filled with all the electrons

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allowed in that level. Chemical bond. Force that holds atom together in a compound. Ion.

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Atomic stability Atoms want to be

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stable. To be stable an atom likes to have 8 electrons flying around it. (except hydrogen and helium) Noble gasses have 8 electrons naturally so they don't make compounds normally.

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Section 1—Stability in Bonding

Why is the Statue of Liberty

Green? Copper, Sulfur, and

Oxygen Some of the matter

around you is in the form of

uncombined element, such as

copper, sulfur, and oxygen.

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-stability in bonding ...
chemical formula. it tells whether
a compound contains and the
exact number of the atoms of
each element in a unit of the
compound. chemical bond. is the
force that holds atoms in a

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compound.

Chapter 19—CHEMICAL BONDS
Block Schedule: 1 session (□ denotes activities recommended for block schedule.) 1. Describe how a compound differs from its component elements. 2. Explain

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what a chemical formula represents. 3. State a reason why chemical bonding occurs.

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