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MCQ Questions for Class 11 Chemistry Chapter 9 Hydrogen ...

Acute pancreatitis remains a clinical challenge, despite an exponential increase in our knowledge of its complex pathophysiological changes. Early fluid therapy is the cornerstone of treatment and is universally recommended; however, there is a lack of consensus regarding the type, rate, amount and end points of fluid replacement.

Fluid resuscitation in acute pancreatitis

But for Cl, the incoming electron goes to the larger 3 rd shell. So, the electronic repulsion is low and hence Cl adds electron more easily than F. Due to the same reason ? eg H of Oxygen is less negative than S. Thus, in modern periodic table, alkali metals have the least -ve ? eg H and halogens have the most -ve ? eg H.

Classification of Salts - A Plus Topper

Chlorine is the chemical name of Cl. Chlorine is commonly used as antiseptic. Visit BYJU'S to understand the properties, atomic mass of chlorine and uses of Chlorine (Cl) explained by the expert teachers.

MCQ Questions for Class 11 Chemistry Chapter 8 Redox ...

Hydrogen Class 11 MCQs Questions with Answers. Question 1. Pure H₂O₂ is : (a) Semi – solid (b) Liquid (c) Solid (d) Gas. Answer: (b) Liquid Explanation: Hydrogen peroxide is the simplest kind of peroxide available (oxygen-oxygen single bond). It is a colourless liquid and is used in aqueous solution for safety reasons.

ML Aggarwal Solutions for Class 9 Maths Chapter 16 ...

Redox Reactions Class 11 MCQs Questions with Answers. Question 1. The oxidation number of Cl in Cl₂O₇ is (a) + 7 (b) + 5 (c) + 3 (d) – 7. Answer: (a) + 7 Explanation: Cl show different oxidation state as -1 to +7 due to vacant d orbital. As oxygen is more electronegative than Cl.

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Classification of Salts What is a salt? A salt is formed in a neutralisation reaction between an acid and a base. Salt is a general term used for the substances that are formed when an acid and a base react with each other. This reaction is called a neutralization reaction. Acid + base ? salt + water [...]

Analysing the Electrolysis of Aqueous Solutions - A Plus ...

ML Aggarwal Solutions for Class 9 Maths Chapter 16 Mensuration is an important study material for students from an exam perspective. Students can refer to the solutions designed by the subject matter experts at BYJU'S having vast conceptual knowledge.

Chlorine (Cl) - Structure, Properties, Uses, and FAQs

Analysing the Electrolysis of Aqueous Solutions. An aqueous solution of a compound is a solution produced when the compound is dissolved in water.; An aqueous solution of a compound contains (a) anions and cations of the compound. (b) hydrogen ions, H⁺ and hydroxide ions, OH⁻ from the partial dissociation of water molecules. During the electrolysis of an aqueous solution of a compound

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