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the radiation emitted from the object and absorbed from the chamber, $I = \frac{1}{4} (T_1^4 - T_2^4)$ (27.4) Wien's law is an empirical equation for the relationship between the temperature of a perfect radiator and the wavelength of the radiation of maximum

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Pacing Guide 1 period Lesson and Problem-Solving Lab

Quantum Theory and the Atom pages 127–134 BLOCK SCHEDULE LESSON PLAN 5.2 Objectives •

Compare the Bohr and quantum mechanical models of the atom. • Explain the impact of de Broglie's wave-particle duality and the Heisenberg uncertainty principle on the modern view of electrons in atoms.

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quantum theory, but the STM is not the only phenomenon with roots in quantum theory. Although the theory explains particle behavior at the micro-

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Answer the following questions regarding the discovery of the atom & its parts. A. Identify 3 conclusions of the gold foil experiment. 1-Atom was more than empty space. 2-There is a dense portion in the center of the atom that contained most of its mass. 3-The nucleus must be positively charged. B.

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