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of a halogen (Br 2, C<sub>2</sub>) with a compound. Homologous series A series of organic compounds that can be described by the same general formula. OR A series of organic compounds in which one member differs from the next with a CH<sub>2</sub> ...

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#### **a NATIONAL SENIOR CERTIFICATE GRADE 12**

4.1.3 Give TWO reasons why they use a catalyst. (2)  
2-methylbut-1-ene (C<sub>5</sub>H<sub>10</sub>) is one of the compounds formed in this reaction. 4.1.4 Write down the structural formula of 2-methylbut-1-ene. (2) 4.1.5 Name the type of reaction that occurs when 2-methylbut-1-ene reacts with hydrogen. (1) 4.2 Consider the structural isomers represented by A, B ...

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$\text{N}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2 \text{NO}(\text{g}) \quad \Delta H = +90,4 \text{ kJ}$  In air, nitric oxide is rapidly oxidised to nitrogen dioxide ( $\text{NO}_2(\text{g})$ ) that initiates the reactions responsible for the formation of smog. Nitrogen dioxide acts as catalyst for the formation of ozone, a key component of smog.

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6.2 Use structural formulae to write a balanced equation for the reaction taking place in the test tube. (5) 6.3 What is the function of the sulphuric acid in the above reaction? (1) 6.4 Why does the method use a water bath instead of direct heating over an open flame? (1) 6.5 State ONE function of the wet paper towel at the top of the test tube.

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