

Olympiad Test Papers CI 6

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The reagent for the test is the equimolar mixture of $ZnCl_2$ and concentrated HCl. the alcohol becomes protonated and the water molecule that is formed leaves, that makes the carbocation, and the nucleophile Cl^- (which is present in excess) readily attacks the carbocation, forming the chloroalkane. Due to the low solubility of the organic ...

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Test tubes, x3; Test tube rack; Test tube holder; Dropping pipettes, x2; Chemicals. Copper(II) sulfate solution, 1.0 M (HARMFUL), about 3 cm³ (see note 3 below) Ammonia solution, 1.0 M, about 10 cm³; Dilute sulfuric acid, 1.0 M (IRRITANT), about 10 cm³; Health, safety and technical notes. Read our standard health and safety guidance. Wear ...

Chlorine (Cl) - Structure, Properties, Uses, and FAQs - BYJUS

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