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Thus the complete nuclear equation is as follows: ${}_{43}^{133}\text{Tc} \rightarrow {}_{44}^{133}\text{C} + {}_{-1}^0\text{e} + \gamma$. The daughter isotope is carbon-12. Test Yourself. Write the nuclear equation that represents the radioactive decay of technetium-133 by beta particle emission and identify the daughter isotope. A gamma ray is emitted

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simultaneously with the beta particle. Answer

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Detecting Nuclear Radiation (page 297)16. Name two devices that are used to detect nuclear radiation.
a. Geiger counters b. Film badges86
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energy. 25.1.2 Describe the three main types of ...

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Chemistry 1110 – Chapter 5 – Nuclear Chemistry – Practice Problems Page | 2 6. Identify the following based on the mass number and atomic number: ${}^0_{-1}e$ 0
A) alpha particle B) beta particle C) positron particle
D) gamma ray E) neutron 7. Gamma rays require the heaviest shielding of all the common types of nuclear radiation

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