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Chlorine (Cl) -
Structure, Properties,
Uses, and FAQs -
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2. By oxidizing iron (II)

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chloride with chlorine
(Cl) $2\text{FeCl}_2 + \text{Cl}_2 \rightarrow$

2FeCl_3 . 3. By oxidizing

iron (II) chloride with

oxygen. $4\text{FeCl}_2 + \text{O}_2$

$+ 4\text{HCl} \rightarrow 4\text{FeCl}_3 + 2\text{H}_2\text{O}$.

Ferric chloride

Test. This test is

conducted to determine

the presence or absence

of phenol in a given

sample. Compounds

such as enols,

hydroxamic acids,

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sulfonic acids, and ...
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Ferric Chloride (FeCl_3
) - Formula, Test,
Colour, Uses, Properties

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the paper industry,
textile industry and in
the construction of
roads; ... How many
 Na^+ and Cl^- ions are
present in a unit cell of
 NaCl ? There will be 4

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ions of each Na^+ and Cl^- . Also, Read: ...

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Uses of Chlorine (Cl) It is used to get rid of the smell of putrefaction; It is used as a disinfectant; Chlorine is used in the treatment of drinking water to kill bacteria; It is used to clean swimming pools; It is

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used in the production
of paper and paper
products; It is used as an
antiseptic; It is used to
produce drugs

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