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NCERT Solutions For Class 10 Hindi Sparsh; NCERT Solutions For Class 10 Hindi Kshitiz; NCERT Solutions For Class 10 Hindi Kritika; ... Hence, the order of depression in freezing point will be $\text{CH}_3\text{COOH} < \text{Cl}_3\text{CCOOH} < \text{F}_3\text{CCOOH}$. 2.32. Calculate the depression in the freezing point of water when 10g of CH_3CH_2

CHClCOOH is added to 250g of water.

NCERT Solutions For Class 12 Chemistry Chapter 2 Solutions

NCERT TEXTBOOK QUESTIONS SOLVED. 3.1. How would you determine the standard electrode potential of the system $\text{Mg}^{2+} + 1 \text{Mg}$? Ans: A cell will be set up consisting of Mg/MgSO_4 (1 M) as one electrode and standard hydrogen electrode Pt, H_2 (1 atm) $|\text{H}^+$ (1 M) as second electrode, measure the EMF of the cell and also note the direction of deflection in the voltmeter.

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