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more easily hydrolysed by aqueous
KOH.

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Give evidence that $[\text{Co}(\text{NH}_3)_5\text{Cl}]\text{SO}_4$ and $[\text{Co}(\text{NH}_3)_5\text{SO}_4]\text{Cl}$ are ionisation isomers. Ans: When dissolved in water, they give different ions in solution which can be tested by adding AgNO_3 solution and BaCl_2

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