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NCERT Solutions CBSE Sample Papers Chemistry Class 12 Chemistry. NCERT IN TEXT QUESTIONS 7.1. Why are pentahalides more covalent than trihalidcs? Ans: The group 15 elements have 5 e-1 s in their valence shell. It is difficult to lose 3e-1 s to form E 3+ and even more difficult to lose 5e-1 s to form E 5+.Thus, they have very little tendency to form ionic compounds.

Chlorine (Cl) - Structure, Properties, Uses, and FAQs

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