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Structure. ... Chemical properties. In the haloalkane molecule, since the CX (X = Cl, Br, I) bond is a polar covalent bond, it is relatively easy to break, so that the haloalkane can undergo various reactions. CF bonds generally have considerable stability, and their reactions differ from ...

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due to large buoyant force acting on it.

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12 Chapter 10 Haloalkanes and
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hydrogen atom(s) in hydrocarbon,*

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aliphatic or aromatic, by halogen atom(s) results in the formation of alkyl halide (haloalkane) and aryl halide (haloarene),

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... Halogenated compounds
persist in the environment due to
their resistance to breakdown by
soil bacteria. 2015-16. Chemistry
282 Haloalkanes and haloarenes
may be classified as follows:
These may be classified as mono,*

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*di, or polyhalogen (tri-,tetra-, etc.)
... by heating a solution of alcohol
in concentrated ...*

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