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What is the Hybridization of Beryllium Dichloride? To know about the hybridization of BeCl_2 (Beryllium Dichloride) we have to take a closer look at the central atom which is Be. Its electronic configuration is $1s^2, 2s^2$, where two electrons are present in the valence shell. During the formation of BeCl_2 , beryllium atom bonds with two chlorine atoms via single covalent bonds.

Kahoot

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Hybridization of BeCl_2 - Hybridization of Be and Cl in BeCl_2 - BYJUS

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