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Building Blocks of Matter MIXED  
REVIEW SHORT ANSWER Answer  
the following questions in the space  
provided. 1. The element boron, B,  
has an atomic mass of 10.81 amu  
according to the periodic table.

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CHAPTER 6 REVIEW Chemical

Bonding SECTION 2 SHORT

ANSWER Answer the following

questions in the space provided. 1.

Use the concept of potential energy to

describe how a covalent bond forms

between two atoms. As the atoms

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involved in the formation of a covalent bond approach each other, the

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## CHAPTER 7 □ HOMEWORK 7-1

VOCABULARY 1. false 2. true 3. false  
4. false 5. true SKILL BUILDER

$\text{AgCl}$  □ In the compound, there is one atom of silver for each atom of chlorine.  $\text{Ba}(\text{OH})_2$  □ The subscript 2 refers to everything inside the parentheses, so there are two

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CHAPTER 9 REVIEW Stoichiometry

MIXED REVIEW SHORT ANSWER

Answer the following questions in the space provided. 1. Given the following equation:  $C_3H_4(g) + xO_2(g) \rightarrow 3CO_2(g) + 2H_2O(g)$  a. What is the value of the coefficient x in this equation?

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40.07 g/mol b. What is the molar mass  
of  $C_3H_4$ ? 2 mol O 2:1 mol H  $2O$  c.  
What is the mole ratio ...

7 Chemical Formulas and Chemical  
Compounds

CHAPTER 3 REVIEW Atoms: The  
Building Blocks of Matter SECTION 3

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SHORT ANSWER Answer the following questions in the space provided. 1. Explain the difference between the mass number and the atomic number of a nuclide. Mass number is the total number of protons and neutrons in the nucleus of an isotope.

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6 Chemical Bonding

CHAPTER 10 REVIEW States of  
Matter SECTION 3 SHORT ANSWER

Answer the following questions in the  
space provided. 1. Match description  
on the right to the correct crystal type  
on the left. b ionic crystal (a) has

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mobile electrons in the crystal c  
covalent molecular crystal (b) is hard,  
brittle, and nonconducting a metallic  
crystal (c) typically has the lowest  
melting point of the four

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Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many

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sulfate ions can be attached to the iron atom. (c) the charge on each Fe ion.

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## HOMEWORK 7-2

the element. The modern form of the periodic table results in the (11) which states that when elements are arranged according to increasing atomic number, there is' a periodic

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repetition of their chemical and  
physical (12) Study Guide for Content  
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