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SHORT ANSWER Answer the following questions in the space provided. 1. Given the following

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equation: $C_3H_4(g) + xO_2(g) \rightarrow 3CO_2(g) + 2H_2O(g)$ 4 a. What is the value of the coefficient
equation? 40.07 g/mol b. What is the molar mass of C_3H_4 ? 2 mol O_2 :1 mol H_2O c. What is
ratio of O_2 to H_2O in the above equation? 0.20 ...

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SECTION 3 continued 4. Table 6on page 485 of the text lists several amphoteric species, but
other than water is neutral. NH_3 a. Identify that neutral compound. b. Write two equations to
demonstrate this compound's amphoteric properties. (Answers will vary, but one equation should show
 NH_3

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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter SECTION 2 SHORT ANSWER Answer the
following questions in the space provided. 1. In cathode-ray tubes, the cathode ray is emitted from the
negative electrode, which is called the

14 Acids and Bases - Password

CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions

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in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain elements in the ...

7 Chemical Formulas and Chemical Compounds

Modern Chemistry 83 States Of Matter CHAPTER 10 REVIEW States of Matter SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. _____ Liquids possess all the following properties except (a) relatively low density.

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Modern Chemistry 1 Chemical Equations and Reactions CHAPTER 8 REVIEW Chemical Equation Reactions Teacher Notes and Answers Chapter 8 SECTION 1 SHORT ANSWER 1. a. d b. a c. b d f. c 2. 8,4,9 3. a. 12 atoms b. 16 atoms c. 51 atoms d. 3 1024 atoms

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Modern Chemistry 1 Solutions CHAPTER 12 REVIEW Solutions Teacher Notes and Answers Chapter 12 SECTION 1 SHORT ANSWER 1. c 2. a 3. b 2. a. alcohol b. water c. the gels 3. The mixture is a colloid. The properties are consistent with those reported in Table 3 on page 404 of the text. The particle size is small, but not too small, and the mixture

3 Atoms: The Building Blocks of Matter - Ms. Sheehan's ...

Modern Chemistry Chapter 15 Acid-Base Titration & pH Section 1 self-ionization of water occurs when two water molecules produce a hydronium (H_3O^+) and a hydroxide (OH^-) ion $2 \text{H}_2\text{O} \rightleftharpoons \text{H}_3\text{O}^+ + \text{OH}^-$. The ionization constant (K_w) of water is: $K_w = [\text{H}_3\text{O}^+][\text{OH}^-] = 1.0 \times 10^{-14} \text{ M}$ Acidic, Basic, & Neutral IF $[\text{H}_3\text{O}^+] > [\text{OH}^-]$ then solution is acidic.

Modern Chemistry Chapter 5 Review Answers Section 1

Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions are attached to the iron atom. (c) the charge on each Fe ion.

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Modern Chemistry Chapter 10 2 Review Answers

CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. (a) has mobile electrons in the crystal (b) is hard, brittle, nonconducting a metallic crystal (c) typically has the lowest melting point of the four

Modern Chemistry Section 1 Review Answers

SECTION 1 Date CHAPTER 11 REVIEW Gases Class SHORT ANSWER Answer the following questions in the space provided. b Pressure — orce For a constant force, when the surface area is doubled, the surface area pressure is (a) doubled. as much. (c) tripled. 7-0 (d) unchanged. Rank the following pressures in increasing order. (c) 76 torr (a) 50 kPa 0 ...

Modern Chemistry Chapter 15 Acid-Base Titration & pH

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