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7 Chemical Formulas
and Chemical
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CHAPTER 5 REVIEW

The Periodic Law

SECTION 1 SHORT

ANSWER Answer the

following questions in

the space provided. 1. c

In the modern periodic

table, elements are

ordered (a) according to

decreasing atomic mass.

(b) according to

Mendeleev ' s original

design. (c) according to

increasing atomic

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number. (d) based on
when they were
discovered.

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ANSWER Answer the
following questions in
the space provided. 1. c
In a Stock system name

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such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom. (c) the charge on each Fe ion.

2 Measurements and
Calculations
CHAPTER 6 REVIEW
Chemical Bonding

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SECTION 4 SHORT
ANSWER Answer the

following questions in
the space provided. 1. b

In metals, the valence
electrons are considered
to be (a) attached to
particular positive ions.
(c) immobile. (b) shared
by all surrounding
atoms. (d) involved in
covalent bonds.

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CHAPTER 7 REVIEW
Chemical Formulas and

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Chemical Compounds

MIXED REVIEW

SHORT ANSWER

Answer the following questions in the space provided. 1. Write

formulas for the following compounds:

CuCO_3 a. copper(II)

carbonate Na_2SO_3 b.

sodium sulfite (NH_4)

3PO_4 c. ammonium

phosphate SnS_2 d.

tin(IV) sulfide HNO_2 e.

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nitrous acid 2.

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6 Chemical Bonding

CHAPTER 3 REVIEW

Atoms: The Building
Blocks of Matter

MIXED REVIEW

SHORT ANSWER

Answer the following questions in the space provided. 1. The element boron, B, has an atomic mass of 10.81 amu according to the periodic table.

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CHAPTER 2 REVIEW
Measurements and
Calculations SECTION
1 SHORT ANSWER

Answer the following questions in the space provided. 1. Determine whether each of the following is an example of observation and data, a theory, a hypothesis, a control, or a model. observation and data a. A research

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team records the rainfall
in inches

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CHAPTER 1 REVIEW

Matter and Change

MIXED REVIEW

SHORT ANSWER

Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance.

homogeneous a. sugar

homogeneous d. plastic

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wrap homogeneous b.
iron filings

heterogeneous e. cement

sidewalk heterogeneous

c. granola bar 2. For

each type of

investigation, select the

most appropriate ...

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Periodic Law Answer
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CH 1 Reading

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Matter & Energy

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14 Section Review

16.1-16.3. White Board

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CHAPTER REVIEW

To give students
practice under more
realistic testing
conditions, give them 60
minutes to answer all of
the ques- tions in this

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Standardized Test
Preparatio Answer the
following items on a
separate piece of paper.
MULTIPLE CHOICE

I. In the modern
periodic table ...

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chemical and physical
properties of elements
are the periodic function
of their atomic numbers
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