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The coefficients in a chemical equation represent the (a) masses in grams of all  
reactants and products. ... MIXED REVIEWcontinued c. If 0.1 mol of N<sub>2</sub> combine with H<sub>2</sub>,  
what must be true about ...

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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the  
following questions in the space provided. 1. Match description on the right to the  
correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c  
covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c)  
typically has the lowest melting point of the four

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7 Chemical Formulas and Chemical Compounds

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11 Molecular Composition of Gases - Madison Public Schools

CHAPTER 5 REVIEW The Periodic Law SECTION 1 ... 38 THE PERIODIC LAW MODERN CHEMISTRY ... CHAPTER 5 REVIEW The Periodic Law MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Consider the neutral atom with 53 protons and 74 neutrons to answer the following questions. 53 a.

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7 Chemical Formulas and Chemical Compounds

9 atoms d.  $\text{Fe}(\text{ClO}_3)_2$  5. N  $2\text{O}_5$  a. What is the formula for the compound dinitrogen pentoxide? iron(II) oxide b. What is the Stock system name for the compound  $\text{FeO}$ ?  $\text{H}_2\text{SO}_3$  c. What is the formula for sulfurous acid? phosphoric acid d. What is the name for the acid  $\text{H}_3\text{PO}_4$ ? MODERN CHEMISTRY CHEMICAL FORMULAS AND CHEMICAL COMPOUNDS53

5 The Periodic Law

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Write

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formulas for the following compounds: CuCO<sub>3</sub> a. copper(II) carbonate Na<sub>2</sub>SO<sub>3</sub> b. sodium sulfite (NH<sub>4</sub>)<sub>3</sub>PO<sub>4</sub> c. ammonium phosphate SnS<sub>2</sub> d. tin(IV) sulfide HNO<sub>2</sub> e. nitrous acid

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