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7 Chemical Formulas and Chemical Compounds

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Modern Chemistry . Name Section Quiz, continued Class Date 6. How many atoms are present in one formula unit of barium acetate, c. 15 d. 16 7. What is the formula for the compound dinitrogen tetroxide? a. d. NO 8. The term acid usually refers to a. a solution of the acid compound in water.

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Modern Chemistry 57 Chemical Bonding CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Label each of the following statements as True or False: \_\_\_\_ a. If the formula mass of one molecule is x u, the molar mass is x g/mol. \_\_\_\_ b.

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Modern Chemistry • CHAPTER 7 HOMEWORK 7-2 ... CHAPTER 7 • HOMEWORK 7-2 VOCABULARY 1. d 2. c 3. e 4. a 5. g 6. h 7. b 8. f GRAPHIC ORGANIZER Name of Compound or Ion Formula of Compound Copper(II) ion Cu<sup>2+</sup> Ammonium ion NH<sub>4</sub><sup>+</sup> Bismuth(III) bromide BiBr<sub>3</sub> Stannous fluoride SnF<sub>2</sub> Magnesium fluoride MgF<sub>2</sub> Iron(III) cyanide Fe(CN)<sub>3</sub>

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER Answer the following questions in the space provided Modern chemistry chapter 7 review answers section 4. 1. c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom.

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HOMEWORK 7-2

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO<sub>3</sub> a. copper(II) carbonate Na<sub>2</sub>SO<sub>3</sub> b. sodium sulfite (NH<sub>4</sub>)<sub>3</sub>PO<sub>4</sub> c. ammonium phosphate SnS<sub>2</sub> d. tin(IV) sulfide HNO<sub>2</sub> e. nitrous acid 2.

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