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6 STUDY GUIDE

Chemical Bonding

SECTION 2

COVALENT BONDING

AND MOLECULAR

COMPOUNDS SHORT

ANSWER Answer

the following

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questions in the
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1. Use the
concept of
potential energy
to describe how
a covalent bond
forms

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is a working
presentation of
the notes for
this chapter.
Meaning that we
may or may not
cover all of the
material here.

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chemical
bonding. a
mutual
electrical
attraction
between the

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nuclei and the
valence
electrons of
different atoms

...

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Chemical Bonding
Chapter Exam

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Choose your
answers to the
questions and
click 'Next' to
see the next set
of questions.

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The Chemical Bonding chapter of this Holt McDougal Modern Chemistry Companion Course helps students learn the essential lessons associated with chemical bonding.

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chemical
compounds tend
to form so that
each atom, by
gaining, losing
or sharing
electrons, has
eight electrons
in its highest

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occupied energy
level ... Modern

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Terms.

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39 Terms.

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Chapter 6 Notes

- Chemical

Bonding Chemical

bond - A mutual

electrical

attraction

between the

nuclei and

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valence
Chemical Bonding

Test Answers
different atoms
that binds the
atoms together

6-1 Introduction
to Chemical
Bonding I. Types
of Chemical

Bonding A. Ionic
Bonding 1.

Chemical bonding
that results
from the

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electrical
attraction
between large

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6; Chemical
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...

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Chemical Bond A
link between

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atoms that
results from the
mutual
attraction of
their nuclei for
electrons

Electrostatic
attraction
between proton
and electron

Classified by
the way the
valence e⁻ are
distributed

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around nuclei of
combined atoms

Types of Bonds

Ionic A chemical bond resulting from electrostatic attraction between cations and anions In a ...

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definition. high potential energy to a lower potential energy. A: Types of Chemical Bonding. valence electrons. cations. anions. ionic bonding - definition. giving up one or more electrons. opposite

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Chapter 6
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attraction of
the ions is what
forms the bond.
covalent bonding

...

Chemical bonding
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CHAPTER 6 REVIEW
Chemical Bonding
SECTION 1 SHORT
ANSWER Answer
the following

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questions in the
space provided.

1. a A chemical bond between atoms results from the attraction between the valence electrons and of different atoms.
(a) nuclei (c) isotopes (b) inner electrons

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(d) Lewis
structures 2. b

A covalent bond
consists of (a)
a shared
electron.

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all the
potential
multiple choice
questions that a
student taking
Mr. New's

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as a review.

Teachers, please
feel free to use
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classes.

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