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Section 5.1: The History of the Periodic Table Dmitri Mendeleev (1869) - first person to organize the elements in a chart Organized about 70 elements by increasing atomic mass Left blank spaces for elements which were not discovered yet

Modern Chemistry Chapter 5 Review The Periodic Law Section ...

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Chapter 5 - The Periodic Law

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a. 2 and 4 c. 4 and 5 b. 3 and 4 d. 5 and 6 ____ 21. The discovery of the noble gases changed Mendeleev's periodic table by adding a new. a. period. c. group. b. series. d. sublevel block. ____ 22. In the modern periodic table, elements are ordered according to. a. decreasing atomic mass. c. increasing atomic number. b. Mendeleev's original ...

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5 The Periodic Law

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CHAPTER 5 The Periodic Law - mchsapchemistry.com

CHAPTER 6 REVIEW Chemical Bonding SECTION 5 SHORT ANSWER Answer the following questions in the space provided. 1. Identify the major assumption of the VSEPR theory, which is used to predict the shape of atoms. Pairs of valence electrons repel one another. 2. In water, two hydrogen atoms are bonded to one oxygen atom. Why isn't water a linear ...

Modern Chemistry Chapter 5 The Periodic Law Section 1 Answers

Modern Chemistry 2 The Periodic Law CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. ____ In the modern periodic table, elements are ordered (a) according to decreasing atomic mass.

10 States of Matter - Ms. Agostine's Chemistry Page

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6 Chemical Bonding

Chapter 5 - The Periodic Law 5-1 History of the Periodic Table I. Mendeleev's Periodic Table A. Organization 1. Vertical columns in atomic weight order a. Mendeleev made some exceptions to place elements in rows with similar properties (1) Tellurium and iodine's places were switched 2. Horizontal rows have similar chemical properties

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CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

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Holt modern chemistry chapter 5 section 2 review answers

CHAPTER 10 REVIEW States of Matter SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. Identify whether the descriptions below describe an ideal gas or a real gas. ideal gas a. ... SECTION 5 MODERN CHEMISTRY STATES OF MATTER. STATES OF MATTER MODERN CHEMISTRY.

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