

## Modern Chemistry Chapter 11 Review

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CHAPTER 11 REVIEW Molecular Composition of Gases MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. The average speed of a gas molecule is most directly related to the . (a) polarity of the molecule (b) pressure of the gas (c) temperature of the gas (d) number of moles in the sample 2.

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### 6 Chemical Bonding

CH 1 Reading Assignment Modern Chemistry. CH 1 Vocabulary-New. CH 1  
Mixed Questions. CH 1 Matter & Energy Vocabulary-New ... Nuclear  
Chemistry Review. Nuclear Chemistry Crossword Puzzle . Unit 1 Review  
Questions. ... Chapter 11 Physical Chapter 11 Molar Volume CH 11 Gases  
& Gas Laws.

### Modern Chemistry Chapter 11 Section 3 Review Answers

Review Previous Concepts Gases CHAPTER 11 1. What is a mole? What  
information does the molar mass of a substance relate about the  
substance? 2. What is the kinetic-molecular theory of matter? Key  
Concepts LOUISIANA STANDARDS LA.PS.29 Predict the properties of a gas  
based on gas laws. (e.g., temperature, pressure, volume). (PS-H-C7)  
340 CHAPTER 11

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You can test your readiness to proceed by answering the Review Questions at the end of the chapter. This might also be a good time to read the Chapter Objectives, which precede the Review Questions. ...

414 Chapter 11 Modern Atomic Theory. 11.1 The Mysterious Electron 415

When a guitar string is plucked, the string vibrates up and down in a

...

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the rates of effusion of gases at the same temperature and pressure are inversely proportional to the square roots of their molar masses

Rate of effusion of A / B =  $\sqrt{M_b} / \sqrt{M_a}$

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