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CHAPTER 11

REVIEW Molecular
Composition of
Gases MIXED

REVIEW SHORT

ANSWER Answer

the following
questions in the
space provided.

1. The average
speed of a gas
molecule is most
directly related

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to the . (a)
polarity of the
molecule (b)
pressure of the
gas (c)
temperature of
the gas (d)
number of moles
in the sample 2.

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The Gases
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your readiness
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Review Previous
Concepts Gases
CHAPTER 11 1.
What is a mole?

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What information does the molar mass of a substance relate about the substance? 2.

What is the kinetic-molecular theory of matter? Key

Concepts

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STANDARDS

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the properties
of a gas based
on gas laws.
(e.g.,
temperature,
pressure,
volume). (PS-H-
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11

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CHAPTER 6 REVIEW

Chemical Bonding

MIXED REVIEW

SHORT ANSWER

Answer the
following
questions in the
space provided.

1. a.

**6 Chemical
Bonding**

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the rates of effusion of gases at the same temperature and pressure are inversely proportional to the square roots of their molar masses

Rate of effusion of A / B = $\sqrt{M_b} / \sqrt{M_a}$

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