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Chapter Test Chapter: States of Matter PART I In the space provided,

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write the letter of the term or phrase that best completes each statement or best answers each question.

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SECTION 1 continued 6. Some binary compounds are ionic, others are covalent. The type of bond favored partially depends on the position of the elements in the periodic table.

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Modern Chemistry 71 Chapter Test Name Class Date Chapter Test B, continued 17. A substance combines with oxygen, releasing a large amount of energy as heat and light, in a(n) . 18. The decomposition of a substance by an electric current is called . 19. A(n) orders the elements by the ease with which they undergo certain chemical reactions. 20.

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Modern Chemistry 130 Chapter Test Name Class Date Chapter Test B, continued PART II Write the correct term (or terms) in the space provided. 9. A substance that ionizes almost completely in aqueous solutions, producing H^+ ions, is a(n) acid. 10. An acid that contains hydrogen and only one other element is called a(n) acid. 11.

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Modern Chemistry Chapter 7 test. STUDY. PLAY. The basis of an ionic bond is the__ absorption of water into their solid structures. Elements tend to gain or lose electrons in order to acquire the electron configuration of a__ noble gas. Ions that are formed from more than one atom are called__

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Chapter 1 an Introduction to Chemistry 3 I would watch the buds swell in spring, the mica glint in the granite, my own hands, and I would say to myself: "I will understand this, too Modern chemistry chapter 4 test answer key. Modern chemistry chapter 4 test answer key

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Modern Chemistry 141 Chapter Test Name Class Date Chapter Test B, continued 18. When a strong acid is titrated with a weak base, the pH

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of the solution at the equivalence point is than 7. 19. A is a highly purified solid used to check the concentration of a standard solution. 20. A 1 M solution of NaOH will have a pH that is

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Modern Chemistry Chapter 3 Test B Answers

Modern Chemistry 30 Chapter Test Name Class Date Chapter Test A, continued _____21. According to Einstein, which of the following can

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behave like a wave and also like a stream of particles? a. a noble gas
b. the atomic nucleus c. electromagnetic radiation d. a hydrogen atom
in the ground state _____22.

7 Chemical Formulas and Chemical Compounds

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continued PART III Match the letter on the diagram with the term for
that point or region. 18. Critical point 19. Vapor 20. Normal freezing
point 21. Triple point 22. Solid 23. Normal boiling point 24. Liquid
25. Critical pressure PART IV Write the answers to the following
questions in the space provided. 26.

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Modern Chemistry Chapter 11. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. jkinsler12. Gases and Gas Laws. Terms in this set (26) Ideal Gas Law. $PV=nRT$. P. Pressure. V. Volume. n. ... temperature and pressure are inversely proportional to the square roots of their molar masses Rate of effusion of A / B = $\sqrt{M_b}$...

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Modern Chemistry 88 Chapter Test Chapter: States of Matter PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. _____ 1. According to the kinetic-molecular theory, particles of an ideal gas a. attract each other but do not collide. b. repel each other and collide.

Assessment Chapter Test B - Ed W. Clark High School

Modern Chemistry 111 Chapter Test Name Class Date Chapter Test B, continued 18. Dissolution processes with negative enthalpies of solution are processes. 19. is the solution process with water as the solvent. 20. Liquid solutes and solvents that are not soluble in each

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other are. 21. A solute molecule that is surrounded by solvent molecules is. 22. $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$

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Modern Chemistry 46 Chapter Test Chapter: Chemical Bonding In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. _____ 1. The charge on an ion is a. always positive.

Modern Chemistry Chapter Test B Answer Key

Assessment Chapter Test B - Baupedia. Modern Chemistry 143 Chapter Test Name Class Date Chapter Test B, continued PART V Write the answers to the following problems on the line to the left, and show your work in the space provided. _____ 39. What is the hydronium ion concentration of an aqueous solution that has a pH of 5.0? _____ 40.

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Holt McDougal Modern Chemistry 3 Chapter Test Chapter Test B, continued 16 Modern chemistry chapter 3 test b answers. The measure of the ability of an atom in a chemical compound to attract electrons from another atom in the compound is called _____. 17. The energy required to remove one electron from an atom is called its _____. 18.

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