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more advanced

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book was written a few decades ago to cover what was the solid state course at Oxford at that time. Some parts of the course have since changed, but other parts are well covered in this book.

- Solid-State

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Luth, Springer-
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Solid State
Physics. 7. find
the correct one
. A. Conversion
from the solid
state to liquid

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state-freezing
B. conversion
from the solid
state to gaseous
state-fusion C.
conversion from
the Solid state
to gaseous
state-
sublimation D.
conversion from
the liquid state
to solid state-
vaporisation .

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Methods of
Physics are also

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A metal
crystallizes
with a face-
centered cubic
lattice. The
edge of the unit
cell is 408 pm.
The diameter of

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Answer**
the metal atom
is [CBSE AIPMT
2012]

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(denoted as CSS;
or Bureaucracy)
is an elite
permanent
bureaucratic
authority, and
the civil
service that is
responsible for
running the
civilian
bureaucratic
operations and
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secretariats and
Answer
directorates of
the Cabinet of
Pakistan.

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Answer

Centered Cubic
(BCC) structure
with atomic
radius 0.123 \AA .
Find the lattice
constant. A) 0
B) 4.587 \AA

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1. The three dimensional graph of lattice points which sets the pattern for the whole lattice is called (a) Space lattice (b)

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Simple lattice
Answer
(c) Crystal
lattice (d) Unit
cell. 2. The
edge of unit
cell of fcc Xe
crystal is 620
pm. The radius
of Xe atom is
(a) 189.37 pm
(b) 209.87

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Unit 4: Solid
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Bank Q. No 1 2 3
4 5 6 7 8

Question Answer

When a P-type
and N-type
semiconductors
are joined
together to form
a PN junction

...

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Question Answers**

...
In a solid 'AB'
having NaCl

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Answer

structure, 'A' atoms occupy the corners of the cubic unit cell. If all the face-centred atoms along one of the axes are removed, then the resultant stoichiometry of the solid is [IIT SCREENING 2001]

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Group B Physics
Chapter Solid
State Physics

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