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Liquid-solid adsorption kinetic data are mostly interpreted by a limited number of rate equations, (pseudo-first, pseudo-second order and Elovich). Besides, a fundamental tool in heterogeneous reactions, the Langmuir theory of adsorption, is used almost exclusively to model equilibrium isotherms but, surprisingly, not kinetics.

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for specifying (a) whether a given lumped kinetic model is kinetically equivalent to the original kinetic model (1) Page 468 May, 1973 or (2), and (b) how the kinetic parameters of the original kinetic system can be evaluated from the available mea- surements of the lumped kinetic system.

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4 CHAPTER 1 The Basics of Reaction Kinetics for Chemical Reaction Engineering The next task in describing a chemically reacting system is the identifica tion of the reactions and their arrangement in a network. The kinetic analysis of the network is then necessary for obtaining information on the rates of individ

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The first model assumes a variable density and constant volume during the pyrolysis process, whereas the second model assumes a constant density and a variable volume. An important feature of these models is that they can accommodate kinetic models for which no analytical representation of the pyrolysis kinetics is available.

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