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solutions are the product of many people including the late John Lamarsh and his students as well as the students at Penn State who used this text. I wish to thank all of them including the graduate assistants who worked with me to develop the course on which this edition is based.

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Chapter 1 Solution 2.14)

The planet earth has a mass of approximately 24.6×10^{22} kg. If the density of the earth were equal to that of nuclei, how big would the earth

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be? 2.16) The fission of the nucleus of ^{235}U releases approximately 200MeV. How much energy (in kilowatt ? hours and megawatt ? days) is released when 1 g of ^{235}U undergoes fission?

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Exercise. questions 5)

The isotopic abundance of naturally occurring hydrogen is as follows:
1H 99.985 atom % 2H 0.015 atom%
When H₂ gas is formed from naturally occurring hydrogen, what percentages of the molecules have molecular weights of approximately 2, 3 and 4? Solution: $P(1,1H_2) =$

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