

## **Inorganic Chemistry Principles Of Structure And Reactivity James E Huheey**

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***Inorganic Chemistry: Principles of Structure and ...*  
While it covers the basics of inorganic chemistry, I felt the early chapters**

**on chemical bonding are rather tedious repeats of general chemistry or belong in physical chemistry class. Instead, a more detailed discussion of bonding in metals/metal complexes is needed, e.g. ligand field theory.**

**Inorganic chemistry; principles of structure and ...**

**The prerequisite courses are General Chemistry I/II, Quantitative Analytical Chemistry, and Organic Chemistry I/II: atomic structure (e.g. atomic numbers, mass numbers, isotopes, subatomic particles), dimensional analysis, periodic table, naming, ions and ionic compounds, stoichiometry, formula weights, interconverting masses and moles, thermochemistry, enthalpy, wave behavior of matter, quantum numbers, quantum mechanics and atomic orbitals, representations of orbitals (s,p,d,f), electron ...**

**Inorganic Chemistry**

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**Inorganic Chemistry by Huheey, James E**

**reviews Inorganic Chemistry: Principles of Structure and Reactivity, 4th ed. James E. Huheey, Ellen A. Keiter, and Richard L. Keiter. Harper Collins: New York, 1993. Figs. and tables. xvii + 1052 pp. 20.0 cm. x 25.3 cm \$72.00. The first edition of this popular text appeared in 1972 and was**

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