

Icse Board 9th Cl Solution

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2NaCl? 2Na(s) + Cl₂(g) Preferential discharge of ions: When there are more than one cations or anion the process of discharge becomes competitive in nature. For example, in the electrolysis of NaCl solution, apart from Na⁺ and Cl⁻ ions the solution of sodium chloride also contains H⁺ and OH⁻ ions due to ionisation of

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Solution: Energy is required to break or make one mole of particular bond in kJ/mol. C-H = 412kJ/mol, Cl-Cl = 242kJ/mol, C-Cl = 331kJ/mol, H-Cl = 432kJ/mol. Step 1: Total energy required to break one C-H and Cl-Cl bond is. 412kJ/mol +241kJ/mol = 654kJ/mol. Step 2: Energy released during the formation of new C-Cl and H-Cl bond is

NCERT English Solutions for Class 9 Beehive Chapter 2 - VEDANTU

Precipitation of heavy metal ions from the aq. solution of their salts : FeCl₃ + 3NH₄OH ? Fe(OH)₃ + 3NH₄Cl Brown ppt. AlCl₃ + 3NH₄OH ? Al(OH)₃ + 3NH₄Cl White ppt. CrCl₃ + 3NH₄OH ? Cr(OH)₃ + 3NH₄Cl Green ppt. Phosphorus: Allotropy of Phosphorus: a) White phosphorus: Translucent white waxy solid. Extremely reactive

Revision notes on P-Block Elements | askIITians

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Meaning, Important Terms, Electrolysis, Redox Reaction - BYJUS

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