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silver chloride and barium nitrate. a. Write and balance the chemical equation. 2
 $\text{AgNO}_3 + \text{BaCl}_2 \rightarrow 2 \text{AgCl} + \text{Ba}(\text{NO}_3)_2$ b. If 39.02 grams of barium chloride are
reacted in an excess of silver nitrate, how many ... Extra Practice - Stoichiometry
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Concept Review: Calculating Quantities in Reactions 1. mole 2. balance 3. relative 4. coefficients 5. molar mass 6. liquids 7. density 8. Avogadro ' s number

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SHORT ANSWER Answer the following questions in the space provided. 1. b The

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coefficients in a chemical equation represent the (a) masses in grams of all reactants and products.

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Problem Type 3: Given is a mass and unknown is an amount in moles. Mass of given substance (g) Reaction Stoichiometry Problems, continued Section 1 Introduction to Stoichiometry Amount of unknown substance (mol) Amount of given substance (mol) Mass of unknown substance (g) Amount of unknown substance (mol) Amount of given substance (mol)

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A balanced chemical equation may be used to describe a reaction ' s stoichiometry (the relationships between amounts of reactants and products). Coefficients from the equation are used to derive ... 7.3: Reaction Stoichiometry - Chemistry LibreTexts

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Section 1 Introduction to Chapter 9 Stoichiometry

Section Quiz: Gas Volumes and the Ideal Gas Law In the space provided, write the letter of the term or phrase that best completes each sentence or best answers each question.

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