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week of studying gases will provide her with answers to the questions her older brothers and sisters posed to her the

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night before at a family dinner. When Ted, who is a mechanic for a Formula One racing team, learned that Lilia was going to be studying gases in her chemistry class, he asked her to find out how to calculate gas density.

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Concept Suppose you want to carry out
a reaction that requires combining one
atom of iron with one atom of sulfur.
How much iron should you use? ... Yes;
the answer has the correct units of
atoms of boron. Is the number of
significant figures correct? Yes; the
mass of boron was given ...

Chapter 13 Gases - An Introduction to
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A pure substance made up of only one type of atom. Matter Anything that takes up space and has mass. Mass A measure of the amount of matter. The study of the composition, structure, and properties of mat... Anything that has mass and takes up space. The physical forms of matter, which are solid, liquid,...

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I ordered two books for my high schooler's Chemistry class, "section reviews" and "study guide" by the same authors, to go with her text book "Modern Chemistry". I was very disappointed to find out that these two books are identical in the first 80 pages, the same title, the same contents, the same page number etc, but the cover is completely ...

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Name Class Date Problem Solving continued
Rearrange the ideal-gas-law equation to solve for the unknown quantity, V . $PV = nRT$ $V = \frac{nRT}{P}$
COMPUTE EVALUATE Are the units correct? Yes; units canceled to give liters of SO_2 . Is the number of significant figures correct?

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Chemistry Live! – Worked Solutions 2
Chapter 9 - The Mole Concept 9.1 Since a sulfur atom ($A_r = 32$) weighs twice as much as an oxygen atom ($A_r = 16$), 32 g of sulfur atoms and 16 g of oxygen atoms both contain the same number of atoms.

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