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of solution.**

**Molality is the  
number of moles**

**of solute per kilogram of solvent. 27.**

**Solutions are homogeneous mixtures.**

**Suspensions and colloids are heterogeneous mixtures.**

**Solutions have the smallest particle size, followed by the**



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**size of colloid  
particles and  
suspension  
particles.**

**Solutions and  
colloids do not  
settle out on  
standing, but ...**

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***solution. a  
homogeneous  
mixture of two or  
more substances  
uniformly  
dispersed  
throughout a  
single phase.***

***suspension. a  
mixture in which***

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***particles of a material are more or less evenly dispersed throughout a liquid or gas. solute.***

***Assessment  
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***solution. a homogeneous mixture of two or more substances uniformly dispersed throughout a single phase. ... a mixture consisting of tiny particles that are intermediate in size between those solutions***

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**and those in  
suspensions and  
that are  
suspended in a  
liquid, solid, or  
gas.**

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by ... a mixture  
consisting of tiny**

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intermediate in  
size between  
those solutions  
and those in  
suspensions and  
that are**

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