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IPC 2nd Semester Exam Review – ANSWER KEY At the surface of the solute, solvent particles surround solute particles (due to +/- attractions) and pull them away into the solution. stirring, increased surface area, high temperature. no stirring/shaking, high pressure, low temperature.

Fall Semester Exam Review Answers - MRS. TYNDALL'S ...

Fall Semester Review – IPC Calcium oxide (CaO) Water ... eventually fall out. \_\_\_\_ A solution of two or more metals. Matching: A. pH B. Base C. Acid D. Neutral \_\_\_\_ pH 7, when there is equal amount of acid and base (distilled water is also this). ... fall semester review.pub Author:

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PDF Fall Semester Review - IPC - winterschemistry.com 3. A metal and non-metal will form a \_\_\_\_\_ compound; two non-metals form a \_\_\_\_\_ compound. Fall Semester Review - IPC Calcium oxide (CaO) Water Find the molecular mass of the following compounds Classify (ionic, covalent or polyatomic) and name the following compounds. 1.

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Fall Semester Review – IPC - Mr. Winters

IPC 1st Semester Exam Review – ANSWER KEY applied. pure. applied. pure. Adding salt makes the water boil at a higher temperature. Adding salt to water does cause it to boil at a higher temperature. Joe correctly tested a single variable and ran repeated trials, but he did not have a control (water without salt) for comparison.

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