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AP Chemistry:

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Electrochemistry Multiple Choice Answers 14. Questions 14-17 The spontaneous reaction that occurs when the cell in the picture operates is as follows: $2\text{Ag}^+ + \text{Cd (s)} \rightarrow 2\text{Ag (s)} + \text{Cd}^{2+}$
(A) Voltage increases. (B) Voltage decreases but remains $>$ zero.

CONCEPT REVIEW – Multiple Choice

APEF – Electrochemistry – Multiple Choice Questions A

1. Which of these chemical reactions is an oxidation-reduction reaction? A. $\text{Fe} + \text{S} \rightarrow \text{FeS}$ C. $\text{AgNO}_3 + \text{NaCl} \rightarrow \text{AgCl} + \text{NaNO}_3$ B. $\text{CO}_2 + \text{H}_2\text{O} \rightarrow \text{H}_2\text{CO}_3$ D. $\text{H}_2\text{SO}_4 + 2\text{NaOH} \rightarrow \text{Na}_2\text{SO}_4 + 2\text{H}_2\text{O}$

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MULTIPLE CHOICE. Choose the
one alternative that best
completes the statement or

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answers the question. 1) The gain of electrons by an element is called_____.

A)oxidation ... Testname: CH
_17_PRACTICE_TEST_ELECTROCHEMISTRY.TST MULTIPLE CHOICE.

Choose the one alternative that best completes the ...

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Electrochemistry Questions and Answers - An electric current is passed through an aqueous solution of the following. Which one shall decompose (a) Urea. ...

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needed for these problems
are given on the inside
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AP Chemistry:

Electrochemistry Multiple
Choice Answers

AP Chemistry-

Electrochemistry. Multiple
Choice. Identify the choice

that best completes the
statement or answers the

question. ____ 1. The half-

reaction that occurs at the
cathode during the

electrolysis of molten

sodium bromide is _____.

Chemistry MCQs for Class 12
with Answers Chapter 3 ...

Electrochemistry Part – A

Questions & Answers 1.

Differentiate between

electrolytic cells and

galvanic cells? (Nov. 2005),

(May 2003) S.No electrolytic

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cells Electrochemical cell 1
Electrical energy is converted into chemical energy. Electrochemical cell is the one, in which chemical energy is converted into electrical energy.

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Multiple Choice Questions 1. In an electrolytic cell the electrode at which the electrons enter the solution is called the _____ ; the chemical change that occurs at this electrode is called _____.

General Chemistry II
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REGENTS PRACTICE - Multiple
Choice 16) In any oxidation-
reduction reaction, the
total number of electrons
gained is a. equal to the
total number of electrons
lost b. less than the total
number of electrons lost c.

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greater than the total
number of electrons lost d.

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MCOs on Electrochemistry with answers, Test: 1, Total Questions: 15. Resistance of a conductivity cell filled with a solution of an electrolyte of concentration 0.1 M is 100 ?.

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