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B) The lattice energy for a solid with $2+$ and $2-$ ions should be two times that for a solid with $1+$ and $1-$ ions. C) MgO has a larger lattice energy than LiF. D) Lattice energy is often defined as the change in energy that occurs when an ionic solid is separated into isolated ions in the gas phase.

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He received his Ph.D. in physical chemistry from Indiana University. Steven D. Gammon is the Provost at the University of Maine Fort Kent.

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