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Hostage was the founding Chemistry Content Advisor for The AP Central Website of the College Board. He is active as a consultant to Board for AP Chemistry and is a member of the exam task force for the Chemistry Olympiad program of the American Chemical Society

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The molality of the solution is the moles of solute per kilogram of solvent. Since water has a density of one kilogram per liter, we can find the total moles by the liters of solution. The molality of the solution is . We can treat this solution as 3L of water containing 4mol of NaCl dissolves into a total of 5 ions.

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AP Chemistry Scoring Guidelines from the 2018 Administration Author: The College Board Subject: AP Chemistry Scoring Guidelines - 2018
Keywords: AP Chemistry Scoring Guidelines from the 2018 administration; exam materials; exam information; teacher resources; score information Created Date: 7/2/2018 4:32:15 PM

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AP* Solution Chemistry Free Response Questions page 4 1994 Discuss the following phenomena in terms of the chemical and physical changes of the substances involved and general principles of chemical and physical change. (a) As the system shown above approaches equilibrium, what change occurs to the volume of water in the container?

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