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answer and write
its letter on
the line. 11.

The fourth
principal energy

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level has a. 4
orbitals. b. 16
orbitals. c. 32
orbitals. 9
orbitals. 12. If
the electron
configuration of
an element is
 $1s^2 2s^2 2p^6 3s^2 3p^5$,
the 116 element
is a. iron. b.
bromine. Core
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chlorine. d.
phosphorus.

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answer and write
its letter on
the line. 1.

Which of the
following tasks
would probably
not be be
assigned to an
... Discuss the
role of
chemistry in one

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of these areas:
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Energy, Medicine
and
Biotechnology,
Agriculture, the
Environment, or
the Universe.

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Date MATTER AND
CHANGE Class

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Date Class IONIC
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AND METALLIC
BONDING Practice
Problems In your
notebook, answer
the following
SECTION 7.1 IONS
For each element
below, state (i)
the number of

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valence
electrons in the
atom, (ii) the
electron dot
structure, and
(iii) the
chemical
symbol(s) for
the most

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conversion

factors from

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Apply the techniques of dimensional analysis to a variety of conversion problems • Solve problems by breaking the solution into steps • Convert complex units, using dimensional

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Vocabulary •
conversion
factor •
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analysis

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Study Tip

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cally. Stick to
your deadlines.
with ChemASAP. If
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students can go
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THEIR ...**

Valence and core
Electrons Name
Date 0 0 Purpose

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To discover the arrangements of electrons within atoms. Complete the table on the handout, filling in the missing atoms. Then answer the questions. 1. How does the number of electrons change as you

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problems -
Answers -**

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CHAPTER 12 Study

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(Resources
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the answer is
not an electron
configuration,
even though it
is) ...

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Languages. Math.

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Chemistry is the

study of the of

matter and the

2. that matter

undergoes. ...

Answer the

following

questions in the

space provided.

28. Match each

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activity below
to one of the
five branches of
chemistry. . . .
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