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For each element below, state (i) the number of valence electrons in the atom, (ii) the electron dot structure, and (iii) the chemical symbol(s) for the most

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(the answer is not  
an electron  
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configuration, even  
though it is) ...

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Chemistry is the study of the of matter and the 2. that matter undergoes. ...

Answer the following questions in the space provided. 28.

Match each activity below to one of the five branches of chemistry. ... 14

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MEASUREMENTS  
AND THEIR ...

Valence and core

Electrons Name

Date O O Purpose

To discover the  
arrangements of  
electrons within



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atoms. Complete the table on the handout, filling in the missing atoms. Then answer the questions. 1. How does the number of electrons change as you move from left to right a period? ... Living By Chemistry Teaching and Classroom

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Choose the best  
answer and write  
its letter on the  
line. 11. The fourth  
principal energy  
level has a. 4  
orbitals. b. 16  
orbitals. c. 32  
orbitals. . 9 orbi s.

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12. If the electron configuration of an element is  $1s^2 2s^2 2p^6 3s^2 3p^5$ , the 116 element is  
a. iron. b. bromine.

Core Teaching  
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chlorine. d.  
phosphorus.

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answer and write  
its letter on the  
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1. Which of the  
following tasks  
would probably not  
be assigned to  
an ... Discuss the  
role of chemistry in  
one of these areas:  
Materials, Energy,  
Medicine and  
Biotechnology,

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Environment, or  
the Universe. ...

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Review Objectives

□ Construct

conversion factors

from equivalent

measurements □

Apply the

techniques of

dimensional

analysis to a

variety of

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conversion  
problems □ Solve  
problems by  
breaking the  
solution into steps  
□ Convert complex  
units, using  
dimensional  
analysis  
Vocabulary □  
conversion factor □  
dimensional  
analysis

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