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Chapter 10 Haloalkanes and Haloarenes - NCERT help

(i) CH_2Cl_2 (ii) CHCl_3 (iii) CCl_4 .

Q10.5: A hydrocarbon C_5H_{10} does not react with chlorine in dark but gives a single monochloro compound $\text{C}_5\text{H}_9\text{Cl}$ in bright sunlight. Identify the hydrocarbon. Q10.6: Write the isomers of the compound having formula $\text{C}_4\text{H}_9\text{Br}$. Q10.7: Write the equations for the preparation of 1-iodobutane from

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education for class 9 10 11 12

These MCQ Questions on Coordination Compounds Class 12 with answers pave for a quick revision of the Chapter thereby helping you to enhance subject knowledge. Have a glance at the MCQ of Chapter 9 Chemistry Class 12 and cross-check your answers during preparation. Question 1. The coordination number of cobalt in $[\text{Co}(\text{en})_3]^{3+}$ is. (a) 3. (b) 4.

Chemistry

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Chapter 9 ...

Class 12 Chemistry. The Solid State. Solutions. Electrochemistry. Chemical Kinetics. Surface Chemistry. Isolation of Elements. The P-Block Elements. The D- & F-Block Elements.

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Chemistry Chapter 10 ...

Topper Package chemistry-XII

Permutation and Combinations
Alkyl Halides & Aryl Halides 51 (c) CHCl_3 is an organic compound (d) AgNO_3 is insoluble in CHCl_3 15. The final product formed by distilling ethyl alcohol with excess of Cl_2 and Ca(OH)_2 is (a) CH_3CHO (b) CCl_3CHO (c) CHCl_2CHO (d) $(\text{C}_2\text{H}_5\text{O})_2\text{C=O}$ 16. The end product (B) in the following sequence of

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Chemistry Chapter 10

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(a) Cl^- (b) Cl^+ (c) AlCl_3 (d) AlCl_4^-
6. Arrange the following in order of their increasing reactivity in nucleophilic substitution reactions CH_3F , CH_3I , CH_3Br , CH_3Cl
7. Carry out the following conversions in not more than two steps : (i) Toluene to benzyl alcohol (ii) Benzyl alcohol to phenylethanenitrile
8.

MCQ Questions for Class 12 Chemistry Chapter 10 ...

Class XII Chapter 10 – Haloalkanes and Haloarenes Chemistry Question 10.2:

Give the IUPAC names of the following compounds: (i) $\text{CH}_3\text{CH}(\text{Cl})\text{CH}(\text{Br})\text{CH}_3$

(ii) $\text{CHF}_2\text{CBrClF}$ (iii) $\text{ClCH}_2\text{C}(\text{CH}_3)_2\text{CH}_2\text{Br}$

(iv) $(\text{CCl}_3)_3\text{CCl}$ (v) $\text{CH}_3\text{C}(\text{p-ClC}_6\text{H}_4)_2\text{CH}(\text{Br})\text{CH}_3$

(vi) $(\text{CH}_3)_3\text{CCH}=\text{CClC}_6\text{H}_4\text{I-p}$

Answer (i) 2-bromo-3-chlorobutane (ii)

1-bromo-1- ...

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MCQ Questions for Class 12 Chemistry
Chapter 1 The Solid ...

Class- XII-CBSE-Chemistry

Haloalkanes and Haloarenes Practice
more on Haloalkanes and Haloarenes

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1. Name the following halides according to IUPAC system and classify them as alkyl, allyl, benzyl (primary, secondary, tertiary), vinyl or aryl halides: (i)

$(\text{CH}_3)_2\text{CHCH}(\text{Cl})\text{CH}_3$ (ii)

$\text{CH}_3\text{CH}_2\text{CH}(\text{CH}_3)\text{CH}(\text{C}_2\text{H}_5)\text{Cl}$

XII_CHEM_CH_10_WS_1_21_22.pdf -

Worksheet 1 Name Roll No ...

Class 12 Chemistry Chapter 10 -
Haloalkanes and Haloarenes Important

Questions - Part 1 Q.1. Name the following halides according to IUPAC system and classify them as alkyl, allyl, benzyl (primary, secondary, tertiary),

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vinyl or aryl halides: (i)
(CH₃)₂CHCH(Cl)CH₃ (ii)
CH₃CH₂CH(CH₃)CH(C₂H₅)Cl (iii)
CH₃CH₂C(CH₃)₂CH₂I (iv)
(CH₃)₃CCCH₂CH(Br)C₆H₅ (v)
CH₃CH(CH₃)CH(Br)CH₃ (vi)
CH₃C(C₂H₅)₂CH₂Br ...

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CH₄ + Cl₂ → CH₃Cl + HCl
CH₃Cl + Cl₂ → CH₂Cl₂ + HCl
CH₂Cl₂ + Cl₂ → CHCl₃ + HCl
CHCl₃ + Cl₂ → CCl₄ + HCl

Noted:-
Iodination is a reversible reaction. So it is carried out by heating alkane in the presence of some oxidizing agent like iodic acid (HIO₃) or nitric acid (HNO₃) or mercuric oxide (HgO) which oxidizes HI formed during the reaction.
CH₄ + I₂ → CH₃I + HI
5HI + HIO₃ → 3H₂O + 3I

Topper Package chemistry-XII Unit

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ALKYL HALIDES ...

Class- XII-CBSE-Chemistry Amines
Practice more on Amines Page - 6
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$C_2H_5NH_2 < (C_2H_5)_3N < (C_2H_5)_2NH$
Again, due to the $-R$ effect of C_6H_5 group, the electron density on the N atom in $C_6H_5NH_2$ is lower than that on the N atom in $C_2H_5NH_2$. Therefore, the basicity of $C_6H_5NH_2$ is lower than that of $C_2H_5NH_2$ i.e.,

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Chemistry Chapter 13

$CH_3CH_2CH_2CH_3 + Cl_2 \xrightarrow{UV \text{ light}}$
 $CH_3CH_2CHClCH_3$ (a)
 $CH_3CH_2CH_2CH_2Cl$ (b) $NaCl + H_2SO_4$ (c)
 Cl_2 gas in dark (d) Cl_2 gas in the presence of iron in dark. Answer.
Answer: (a) Cl_2 /UV light

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Chemistry Chapter 2 Solutions ... the
order of depression in freezing point
will be $\text{CH}_3\text{COOH} < \text{Cl}_3\text{CCOOH} < \text{F}_3\text{CCOOH}$. 2.32. Calculate the
depression in the freezing point of water
when 10g of $\text{CH}_3\text{CH}_2\text{CHClCOOH}$ is
added to 250g of water. $K_a = 1.4 \times 10^{-3}$
 $K_f = 1.86 \text{ K kg mol}^{-1}$. Solution: 2.33.
19.5g of CH_2FCOOH is ...

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students understand the concept very well.

MULTIPLE CHOICE QUESTIONS

(i) CH_2Cl_2 (ii) CHCl_3 (iii) CCl_4

Solution: Question 5. A hydrocarbon C_5H_{10} does not react with chlorine in dark but gives a single monochloro compound $\text{C}_5\text{H}_9\text{Cl}$ in bright sunlight. Identify the hydrocarbon. Solution: A number of structural isomers are possible for molecular formula C_5H_{10} .

Question 10.1: (i) $\text{CHCH}(\text{Cl})\text{CH}_2\text{CH}_2\text{CH}(\text{CH}_3)\text{CH}_2\text{CH}_2\text{I}$

Topper's Package chemistry-XII
Nitrogen Containing

Compounds Permutation and

Combinations 91 (a) $\text{C}_2\text{O}_2\text{C}_2\text{N}-\text{CH}_2\text{O}$

Br (b) $\text{C}_2\text{O}_2\text{C}_2\text{N}_2\text{OCHCl}_2$ (c) $\text{C}_2\text{O}_2\text{C}_2\text{N}$

OCHBr_2 (d) $\text{C}_2\text{O}_2\text{C}_2\text{N}_2\text{OCH}_2\text{ClCH}_2$

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47. $\text{OH} \text{ O} \text{ NH}_2$ (-hydroxy amide) Br / KOH Product; Product of this Hoffmann bromamide reaction is: (a) $\text{Ph} \text{ C} \text{ CH}_3 \text{ O}$ (b) $\text{Ph} \text{ CHO}$ (c) $\text{Ph} \text{ CH} \text{ OH} \text{ NO}_3$ (d) $\text{Ph} - \text{CH} \dots$

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Class XII Chapter 10 – Haloalkanes and Haloarenes Chemistry Page 4 of 40

Question 10.2: Give the IUPAC names of the following compounds: (i)

$\text{CH}_3\text{CH}(\text{Cl})\text{CH}(\text{Br})\text{CH}_3$ (ii)

$\text{CHF}_2\text{CBrClF}$ (iii) $\text{ClCH}_2\text{C}(\text{CH}_3)_2\text{CCH}_2\text{Br}$

(iv) $(\text{CCl}_3)_3\text{CCl}$ (v) $\text{CH}_3\text{C}(\text{p-ClC}_6\text{H}_4)_2\text{CH}(\text{Br})\text{CH}_3$ (vi)

$(\text{CH}_3)_3\text{CCH}=\text{CClC}_6\text{H}_4$ I-p Answer

Class 12 Chemistry Chapter 10

Haloalkanes and Haloarenes ...

Give reason for your answer : $1\frac{1}{2} \text{ 2 2 Ni} / \text{Ni Cu} / \text{Cu} \text{ 0.25 V 0.34 V E E}$ (f)

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Calculate m at 298 K of NH_4OH at infinite dilution, given $(\text{OH}) 174.2 \text{ S cm mol}^{-1}$ $(\text{Cl}) 66.2 \text{ S cm mol}^{-1}$ $(\text{NH}_4\text{Cl}) 130 \text{ S cm mol}^{-1}$ $(1/2) (12)$
HS/XII/Sc/Ch/NC/20 /38 20K

CHAPTER 13 HYDROCARBON

Hydrocarbons are composed of Carbon

...

Sulphur molecule is. (a) diatomic. (b) triatomic. (c) tetratomic. (d) Octatomic. Answer. Answer: (d) Octatomic. We hope the given NCERT MCQ Questions for Class 12 Chemistry Chapter 7 The p-Block Elements with Answers Pdf free download will help you. If you have any queries regarding The p-Block Elements CBSE Class 12 Chemistry MCQs Multiple ...

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On the other hand, in case of CH_2Cl_2 , the resultant of the dipole moments of two $\text{C}\text{-}\text{Cl}$ bonds is strengthened by the resultant of the dipole moments of two $\text{C}\text{-}\text{H}$ bonds. As a result, CH_2Cl_2 has a higher dipole moment of 1.60 D than CHCl_3 i.e., CH_2Cl_2 has the highest dipole moment.

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